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1	Total gaseous mercury concentrations at the Cape Point GAW station and
2	their seasonality
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14	Abstract
15	
16	Total gaseous mercury (TGM) has been measured at the WMO Global Atmosphere Watch
17	(GAW) station at Cape Point, South Africa, since September 1995. Annual medians suggest a
18	small but significant decrease of TGM concentrations from 1.29 ng m ⁻³ in 1996 to 1.19 ng m ⁻³
19	in 2004. Background TGM concentrations at Cape Point show a pronounced seasonal
20	variation with a maximum in January and February and a minimum in Juli, August and
21	September. This seasonal variation is opposite in phase to that of CO at Cape Point, whereas
22	at Mace Head in the northern hemisphere the seasonal variations of TGM and CO are in
23	phase. This behavior is in approximate agreement with the seasonal variation of major
24	mercury sources in both hemispheres, suggesting lesser influence of the seasonality of sinks.
25	The different seasonal variations of TGM concentrations in the northern and southern
26	hemispheres may pose an important constraint on the global models of atmospheric mercury.

28 Introduction

29

30 Mercury, in contrast to other metals, is volatile and its vapor can be transported in air over 31 large distances because of its low reactivity and solubility. After oxidation to less volatile and 32 more soluble compounds it is deposited (e.g. Slemr et al, 1985; Lin et al., 2006). Part of the 33 deposited mercury compounds is then converted to more toxic methyl mercury which 34 bioaccumulates in the aquatic food chain. The high concentration of methyl mercury in the 35 predatory fish pose a serious risk for people and animals which depend on a fish diet (e.g. 36 Mergler et al., 2007; Scheuhammer et al., 2007). Consequently, the research on atmospheric 37 mercury has intensified during the last decade (Lindberg et al., 2007).

38

39 Despite of the substantial progress, the understanding of the global atmospheric cycle is still 40 incomplete (e.g. Lin et al., 2006). Natural and anthropogenic mercury emissions are not well 41 known (e.g. Lindberg et al., 2007). The major knowledge gap, however, concerns the 42 mechanism of the oxidation of highly volatile and almost insoluble gaseous elemental 43 mercury (GEM) into less volatile and more soluble oxidized mercury species and their 44 incorporation into particles (Lin et al., 2006). As the oxidized mercury species are readily 45 removed from the atmosphere by dry and wet deposition (Lin et al., 2006), the rate of the 46 GEM oxidation controls the removal of mercury from the atmosphere. Three major oxidation 47 mechanisms by reactions with O₃, OH (Lin et al, 2006), and Br/BrO (Holmes et al., 2006) 48 have been proposed but their relative importance is still unclear (Lin et al., 2006).

49

50 Long term monitoring of mercury can provide valuable constraints on emissions and on the 51 oxidation mechanism of atmospheric mercury. Mercury is now being monitored at many sites 52 in the northern hemisphere (NH) but, to the best of our knowledge, only two longer term 53 systematic measurements have been reported from the southern hemisphere (SH) so far.
54 Ebinghaus et al. (2002a) measured total gaseous mercury (TGM) at the German Antarctic
55 research station from January 2000 to January 2001. Baker et al. (2002) reported first results
56 of the long term monitoring of TGM at the Cape Point observatory in South Africa covering a
57 period from September 1995 to June 1999. The latter measurements have been continued and
58 here we present the trend and seasonal variation derived from measurements covering almost
59 a 10 year period since September 1995 until the December 2004.

60

61 Experimental

62

63 The Cape Point station (34°21'S, 18°29'E) is operated as one of the World Meteorological 64 Organisation (WMO) Global Atmospheric Watch (GAW) baseline monitoring sites. The 65 station is located at the southern tip of the Cape Peninsula within the Cape Point National 66 Park on top of a peak 230 m above sea level and about 60 km south of Cape Town. The 67 station has been in operation since the end of the 1970s. It receives clean marine air masses 68 for most of the time and has, therefore, been used to study the trends and seasonal variations 69 of CO₂, CO, CH₄, N₂O, O₃ and halocarbons at midlatitudes of the SH (e.g. Brunke et al, 1990; 70 Oltmans et al, 1998).

71

As the primary objective of the TGM monitoring at Cape Point was to determine trends and seasonal variations, only 8 samples were typically taken every other week using a programmable sequential sampler providing about 200 samples per year. The sampling and analysis are described in detail by Baker et al. (2002). The technique measures the total gaseous mercury (TGM) under most conditions and was in excellent agreement with other TGM techniques during the intercomparison at Mace Head (Ebinghaus et al., 1999).

TGM measurements at Cape Point are summarized in Table 1. During three years the measurements cover only a part of the year: September to December in 1995, January to March in 1997, and May, July, August and September in 1998. Because of the seasonal variation discussed later they cannot be regarded as representative for the whole year. The remaining medians for 1996 and 1999 - 2004 suggest a small but significant decreasing trend of TGM concentrations of 0.015 ± 0.004 ng m⁻³ year⁻¹(R² = 0.740, n = 7). Annual medians were used because they are less sensitive to extreme values than averages.

88

89 Fig. 1 shows the annual TGM medians for Cape Point together with the TGM medians at the 90 Neumayer station, Antarctica (Ebinghaus et al., 2002a) and southern hemispheric TGM 91 measurements during ship cruises on the Atlantic Ocean in 1994, 1996, and 1999 - 2001 92 (Temme et al., 2003b). An exact comparison of all these measurements is not possible 93 because of their different spatial and temporal coverage. However, the ship measurements in 94 1996 and 2000 agree approximately with those from Cape Point while the data from the 1994 95 are lower for unknown reasons. The TGM concentrations at the Neumayer station in 96 Antarctica tend to be somewhat lower - possibly because of the mercury depletion events.

97

The seasonal variation of TGM and CO concentrations at Cape Point is shown in Fig. 2. For comparison, seasonal variation of TGM and CO concentrations at Mace Head at the west coast of Ireland in the NH is also shown. Mace Head is one of the ALE/GAGE/AGAGE stations used to study trends and seasonal variations of halocarbons, CO₂, CO, CH₄, H₂, O₃ (Simmonds et al., 2005), and mercury (Ebinghaus et al., 2002b). Only baseline data were taken to eliminate the influence of nearby sources. At Cape Point TGM measurements in air with ²²²Rn concentrations of less than 600 mBq m⁻³ were considered to be of baseline quality

in contrast to continental air masses, which exceed 1200 mBq m⁻³ (Brunke et al., 2004). 105 Without ²²²Rn data, the baseline periods at Mace Head were identified by the NAME 106 107 Lagrangian dispersion model (Manning et al., 2003). To eliminate the year to year variation 108 of background TGM, the monthly means of TGM concentrations were normalized to annual 109 medians. The TGM points in Fig. 2 then represent an average of relative monthly means over the period from March 1999 to December 2004, for which the ²²²Rn measurements were 110 111 available, and the uncertainty bars represent their standard deviation. The TGM points and 112 their uncertainty bars for Mace Head were derived in the same way as those for Cape Point. 113 The absolute monthly means of TGM concentrations can be calculated by using the average annual median TGM concentrations over the period 1999 – 2004 of 1.203 \pm 0.038 ng m^3 at 114 Cape Point and 1.724 ± 0.060 ng m⁻³ at Mace Head, respectively. The CO data points and 115 116 their uncertainty bars represent the baseline monthly averages of CO concentrations with their 117 corresponding standard deviations.

118

119 To the best of our knowledge, the seasonal variation of TGM concentration in Fig. 2 is the 120 first one derived from a long term monitoring in the SH. It shows maximum TGM 121 concentrations in January and February (austral summer) and a minimum during austral 122 winter. Although measurements by Ebinghaus et al. (2002a) and Temme et al. (2003a) in the 123 Antarctic cover a whole year, the seasonal variation of their measurements is obscured by the 124 frequent mercury depletion events starting in July and ending in February. A preliminary 125 analysis of Cape Point data from September 1995 to June 1999 suggested a TGM minimum in March - May and a maximum in June - August (Baker et al., 2002), but no attempt was made 126 127 to filter out the baseline data.

128

In spite of the occasionally large uncertainty bars for the monthly mean TGM concentrations,Fig. 2 shows that the seasonal variation of TGM relative to that of CO is opposite in phase in

131 the SH and in phase in the NH. The amplitude of the seasonal variations of 0.134 at Cape 132 Point appears to be somewhat larger than those for Mace Head (0.097), although this 133 difference may not be significant in view of the large uncertainties of the monthly means. The 134 amplitude of the CO concentrations at Cape Point is 32.2 ppb in absolute and 0.49 in relative 135 terms, at Mace Head 69.4 ppb and 0.58, respectively. The inverse average ratio of relative 136 TGM (0.116) to CO (0.54) amplitudes of 4.7 suggest a correspondingly longer lifetime of 137 TGM relative to that of CO, i.e. about 0.83 yr if the lifetime of CO of 0.18 yr is assumed 138 (Warneck, 1988).

139

140 The seasonal variation of CO concentrations is driven mainly by the seasonal variation of its 141 major sink due to the reaction with OH. The major source of OH radicals is photolysis of O_3 142 leading partly to $O(^{1}D)$ atoms and their subsequent reaction with H₂O. The seasonalities of 143 solar irradiation (i.e. wavelength < 310 nm) and of water vapor concentrations lead to a 144 pronounced seasonal variation of OH with a maximum concentration in summer and a 145 minimum one in winter (Spivakovsky et al., 2000). The CO concentration follows the inverse 146 pattern with a delay of about 2 months. The similar seasonal behavior of TGM and CO in the 147 NH was sometimes taken as a sign in favor of OH driven oxidation for TGM (Bergan and 148 Rodhe, 2001). The opposite behavior in the SH, however, suggests that this may be incorrect.

149

We hypothesize that the seasonal variation of TGM in the SH might be driven predominantly by its emissions. A summary of the currently available information about the seasonality of mercury emissions is shown in Fig. 3. The seasonality and magnitude of oceanic mercury emissions were taken from the results of model by Strode et al. (2006). The emissions from biomass burning were calculated using the seasonal variation of CO emissions given by Duncan et al. (2003) and Hg/CO emission ratios of 1.89 and 1.36 ×10⁻⁷ mol/mol in the southern and northern hemispheres, respectively (Ebinghaus et al., 2007). The anthropogenic 157 mercury emissions for 2000 (Wilson et al., 2006) were allocated to the hemispheres and to 158 combustion and other sources. For the combustion sources the hemispheric seasonal 159 variations given by Rotty (1987) were used to calculate their monthly contributions. The 160 interhemispheric flux was calculated using the seasonal cycle of the interhemispheric 161 exchange time derived by Levin and Hessheimer (1996) from ⁸⁵Kr measurements and the 162 seasonally variable north-south TGM gradient. Because of the higher TGM concentrations in 163 the NH the flux is always negative from the NH perspective and positive for the SH. The total 164 emissions, including the north to south transport, vary from 152 t/yr in May to 296 t/yr in 165 December in the SH, and from 105 t/yr in December to 234 t/yr in May in the NH.

166

167 The emissions considered in Fig. 3 yield a total of 4562 t/yr, representing 69% of the total 168 emissions estimated by Mason and Sheu (2002). The remaining 1605 t/yr are broadly ascribed 169 to land emissions (Mason and Sheu, 2002) including volcanic emissions of 110 - 700 t/yr 170 (Nriagu and Becker, 2003; Pyle and Mather, 2003), primary emissions from mercury deposits 171 of about 500 t/yr (Lindquist et al., 1991), and reemissions of previously deposited mercury. 172 The latter two are known to vary with temperature, solar radiation, and precipitation (Gustin 173 et al., 1997) leading to a seasonal variation with maximum emissions in summer and a 174 minimum in winter. As the contributions of volcanic emissions, primary emissions, and 175 reemissions are not well known, the seasonality of the land emissions cannot be estimated. 176 But it will be in phase with the emissions in the SH and thus increase the seasonal amplitude 177 of the overall emissions there. In the NH, it will be out of phase with the emissions and will 178 thus reduce the amplitude of the emission seasonality.

179

In summary, the seasonality of TGM emissions in Fig. 3 is in reasonable agreement with the baseline TGM observations at Cape Point and Mace Head suggesting a lesser influence of the sink seasonality. Of the three potential sinks due to reactions with OH, O₃, and Br (Lin et al,

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2006; Holmes et al., 2006) the most pronounced seasonality can be expected for OH. The
resemblance of the TGM seasonality to the seasonality of the sources is thus more consistent
with O<sub>3</sub> and Br oxidation of elemental mercury than with oxidation by OH.
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187 Conclusion

188

The measurements of total gaseous mercury (TGM) made at the WMO Global Atmosphere Watch (GAW) station at Cape Point, South Africa, since September 1995 suggest a small but significant decrease of annual median TGM concentrations from 1.29 ng m⁻³ in 1996 to 1.19 ng m⁻³ in 2004.

193

194 Baseline TGM concentrations at Cape Point show a pronounced seasonal variation with a 195 maximum in austral summer and a minimum in austral winter. This seasonal variation is 196 opposite in phase to that of CO at Cape Point, whereas in the NH at Mace Head the seasonal 197 variations of TGM and CO are in phase. This behavior seems to be dominated by the seasonal 198 variation of major mercury sources in both hemispheres, suggesting lesser influence of the 199 sink seasonality. We are aware, that this explanation is speculative in view of the combined 200 uncertainties of the emission estimates. But, if the seasonality observed at Cape Point is 201 confirmed at other sites in the SH, the different seasonalities in the northern and southern 202 hemispheres of TGM concentrations may pose an important constraint on the global models 203 of atmospheric mercury.

204

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206

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211	
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332 Figure Labels

333

Figure 1: TGM concentrations measured at Cape Point, Neumayer station (Antarctica) and onboard ships over the Atlantic Ocean in the southern hemisphere. The concentrations are given in ng/m3 (STP, i.e. 273.2 K, 1013 mbar). The points and bars represent the annual medians (cruise medians for ship cruises) and their 95% confidence intervals.

339 Figure 2: Seasonal variation of TGM (relative monthly averages and their standard deviation,

340 see text) and CO (absolute monthly averages and their standard deviations) at Cape Point and

- 341 Mace Head. Only baseline data are considered in this diagram, see text.
- 342

343 Figure 3: Seasonal variation of the emissions from oceans, biomass burning, anthropogenic

344 activities, and of the interhemispheric flux. Details of calculations are given in the text.

345

346 Tables

347

348 Table 1: Summary of the TGM measurements at Cape Point.

349

Year	Range	Average ±	Median	95%	Number of
	$[ng m^{-3}]$	Standard	$[ng m^{-3}]$	confidence	measurements
		deviation		interval of	
		$[ng m^{-3}]$		median	
				$[ng m^{-3}]$	
1995	1.23 – 3.16	1.518 ± 0.372	1.42	1.34 - 1.54	26
1996	0.93 - 1.88	1.297 ± 0.136	1.29	1.26 – 1.31	212
1997	0.73 - 1.60	1.182 ± 0.174	1.17	1.11 - 1.22	50
1998	0.95 - 1.82	1.308 ± 0.167	1.28	1.24 – 1.37	38
1999	0.78 - 1.61	1.244 ± 0.139	1.25	1.23 – 1.27	163
2000	0.49 - 2.18	1.227 ± 0.174	1.21	1.18 - 1.22	173
2001	0.84 - 6.78	1.452 ± 0.817	1.24	1.21 – 1.28	152
2002	0.92 - 1.64	1.200 ± 0.132	1.18	1.14 - 1.20	143
2003	0.93 - 1.60	1.176 ± 0.110	1.15	1.14 - 1.18	125
2004	0.78 - 2.37	1.261 ± 0.247	1.19	1.17 – 1.22	155

- 351 Figures
- 352
- 353 Fig. 1:



355

357 Fig. 2:



361 Fig. 3:

